

Ethylene and Fruit Ripening

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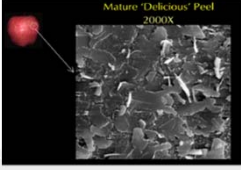





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I. Processes Associated with Fruit Ripening

- Seed maturation
 - ability of seeds to germinate successfully
- Development of wax  skin
- Abscission of fruit



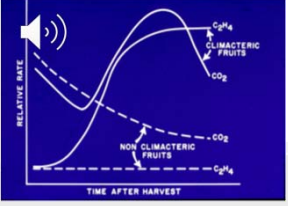


Source: Curry, 2000
<http://postharvest1.trec.wsu.edu/pgDisplay.php?article=PC2000AA#s2>



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I. Processes Associated with Fruit Ripening



- Change in C_2H_4 production (climacteric fruits only)
 - increase in endogenous ethylene concentration until adequate to initiate ripening
 - autocatalytic ethylene production
- Increased sensitivity to C_2H_4 (climacteric fruits only)

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I. Processes Associated with Fruit Ripening


- Change in respiration rate
 - increased O_2 consumption and CO_2 production in climacteric fruits
- Membrane and cell wall changes
 - increased membrane permeability → juice leakage
 - depolymerization of pectins and other cell wall polysaccharides → softening

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I. Processes Associated with Fruit Ripening

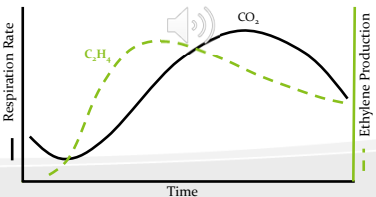


- Protein and nucleic acid changes
 - synthesis of enzymes involved in compositional changes
- Compositional changes 
 - starch to sugar conversion; loss of acids
 - pigment synthesis (anthocyanins & carotenoids) and degradation (chlorophyll)
 - polymerization of tannins and resulting loss of astringency
 - formation of flavor volatiles



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II. Factors Influencing Climacteric Fruit Ripening

- Factors that affect C_2H_4 production and action can all be manipulated to control ripening

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II. Factors Influencing Climacteric Fruit Ripening

- Ethylene treatment
 - promotes faster and more uniform ripening by coordinating the onset of ripening
 - reduced time between harvest and consumption can mean better quality and nutritive value

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Effect of Temperature on Ripening Rate of Stone Fruits Exposed to 100 ppm Ethylene for 48 Hours

Temperature (°C)	Mean Number of Days to Ripen			
	Nectarine	Peach	Plum	Mean
15	6.9	5.6	5.6	6
20	4.3	3.8	3.7	4
25	3.2	2.7	2.7	3

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Ethylene Effects on Fruit Ripening as Indicated by Flesh Firmness of Stone Fruits

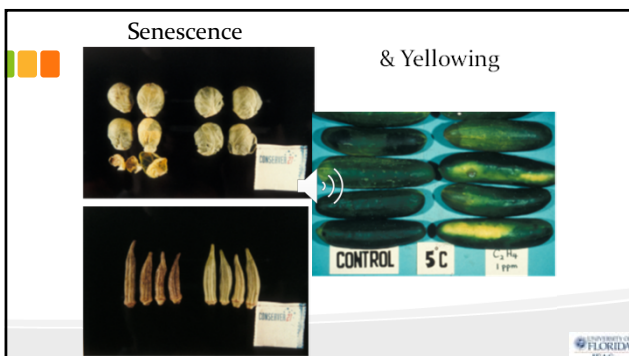
Days at 20°C	Treatment	Flesh Firmness (kg)		
		Nectarine	Peach	Plum
0	At harvest	5.3 ± 1.0	6.9 ± 0.7	2.8 ± 0.9
4	Without added ethylene	1.0 ± 0.5	1.3 ± 0.5	1.7 ± 1.0
4	With 20 ppm ethylene	0.8 ± 0.2	1.0 ± 0.3	0.8 ± 0.5

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III. Undesirable Ethylene Effects

- Undesired ripening and softening of fruits in storage
- Accelerated senescence and loss of green color in leafy and immature fruit vegetables (e.g., cucumbers)
- Abscission of leaves (e.g., cauliflower, cabbage, foliage plants, etc.)
- Sprouting (stimulation or retardation)

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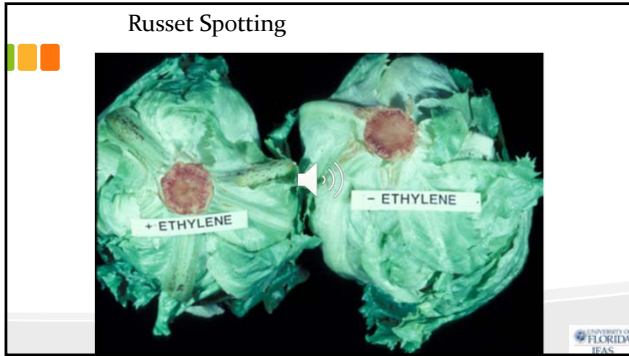


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III. Undesirable Ethylene Effects (cont.)

- Induction of phenolic synthesis
 - bitter principle (isochlorogenic acid) in carrot roots
 - toxic ipomeamarone in sweet potato roots
 - russet spotting on lettuce
 - lignification of asparagus

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III. Undesirable Ethylene Effects (cont.)

- Physiological disorders of ornamental crops
 - 'sleepiness' of carnations (failure of bloom to open)
 - flower and leaf abscission
 - inhibition of shoot and root elongation; gummosis; bud necrosis and flower bud blasting in bulb crops

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III. Undesirable Ethylene Effects (cont.)

- The ultimate undesirable ethylene effect!!

Explosion, fire destroy banana ripening facility

The Packer July 12, 1999

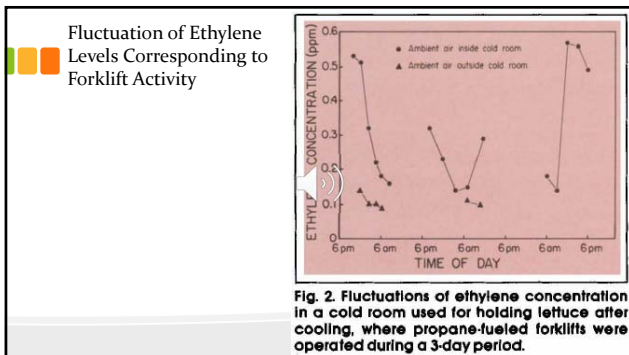
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IV. Postharvest Sources and Levels of Ethylene in Lettuce Handling

Sample locations	Ethylene concentration (ppm)		No. of samples analyzed	Potential sources
	Range	Mean		
Field	A* trace-0.12	—	21	Air pollution
Field to cooler	B* 0.03-0.11	0.07	3	Mechanically injured lettuce, Exhaust from truck, other pollution
Holding areas prior to vacuum cooling	A 0.01-0.61	0.05	47	Exhaust from trucks and forklifts
B 0.01-0.80	0.16	12	Exhaust from trucks, other pollution	
Immediate cooling following cooling	B 0.01-0.29	0.12	11	Vacuum cooling removes much of the C ₂ H ₄ inside cartons
Cold storage rooms of vacuum coolers	A 0.01-2.78	0.33	144	Exhaust from forklifts, other commodities
B 0.01-1.56	0.32	73	Exhaust from forklifts, other commodities	
Inside rail cars at destination	A 0.01-0.19	0.06	14	Decay, other pollution sources
B 0.01-0.02	0.01	3	Decay, other pollution sources	
Inside truck units at destination	A 0.04-0.22	0.08	9	Decay, other pollution sources
B 0.08-0.11	0.09	4	Decay, other pollution sources	
Distribution centers and warehouses	A 0.03-2.49	0.25	22	Exhaust, other commodities
B 0.01-0.78	0.08	42	Exhaust, other commodities	
Retail storage areas	A 0.02-2.95	0.36	19	Other commodities
B 0.06-2.66	0.41	18	Other commodities	
Home refrigerator	A 0.02-1.58	0.25	33	Other commodities

*A = atmosphere external to carton; B = inside carton
Ethac = less than 0.10 ppm

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V. Commercial Use of Ethylene

- Methods of application
 - Cylinders of ethylene or banana gas (C₂H₄ in CO₂) with flowmeters

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V. Commercial Use of Ethylene

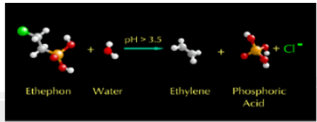
- Methods of application
 - Ethylene generators (liquid ethanol plus catalyst => C₂H₄)



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V. Commercial Use of Ethylene

- Methods of application
 - Ethylene-releasing chemicals (e.g., Ethephon = 2-chloroethane-phosphonic acid)
 - Intended for field application, but sometimes used postharvest



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V. Commercial Use of Ethylene (cont.)

- Ethylene concentration and duration of treatment
 - physiological responses are saturated at 100 ppm
 - mature climacteric fruit should initiate endogenous ethylene production within no more than 72 hours
 - degreening should continue for no more than 72 hours or risk increased peel senescence and decay

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V. Commercial Use of Ethylene (cont.)

- Used for ripening of climacteric fruits
 - Bananas
 - Tomatoes
 - Avocados

} Very common use

 - Mangoes
 - Papayas
 - Persimmons
 - Casaba & Honeydew melons

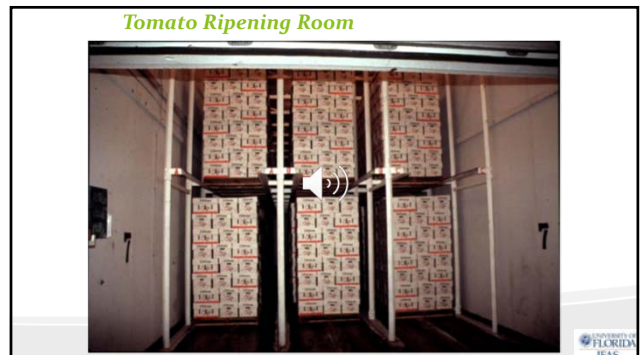
} Not common

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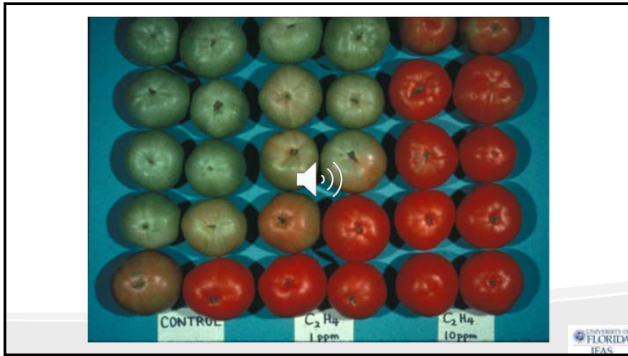
V. Commercial Use of Ethylene (cont.)

- Ripening climacteric fruits
- Recommended conditions (tomatoes)
 - 20 to 21°C
 - 90 to 95% RH
 - 100 to 150 ppm C₂H₄
 - Air circulation = 20-40 ft³ per minute per ton of product
 - Ventilation = 1 air change per 6 hours or open room for 0.5 h twice per day (prevent CO₂ build-up)

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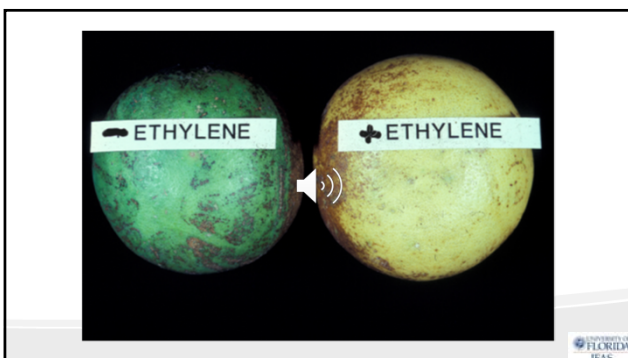
V. Commercial Use of Ethylene (cont.)

- Degreening of citrus fruits
- Recommended conditions (Florida)
 - 28 to 29°C
 - 90 to 96% RH
 - 5 ppm C₂H₄
 - Air circulation = 10 ft³ per min. per box
 - Ventilation = 1 air change per hour or sufficient to maintain <0.2% CO₂

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VI. Avoiding Exposure to Ethylene

- Exclusion of ethylene from storage rooms
 - use of electric forklifts
 - C₂H₄ absorber on forklift exhaust
 - avoiding other pollution sources
 - avoiding mixing ethylene-producing and ethylene-sensitive crops

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VI. Avoiding Exposure to Ethylene (cont.)

- Removal of ethylene from storage rooms
 - use of adequate ventilation (air exchange)
 - use of ethylene absorbers
 - Potassium permanganate (alkaline $KMnO_4$ on inert pellets = "Ethylene Control", etc.)
 - Activated and brominated charcoal +/- $KMnO_4$ = "Stayfresh" absorbers
 - Palladium-impregnated zeolite (It's Fresh! - the most effective)

Source: Ethylene Control, Inc.

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VI. Avoiding Exposure to Ethylene (cont.)

- Removal of ethylene from storage rooms
 - use of ozone or UV radiation to oxidize ethylene:
 - $O_3 + UV \rightarrow O_2$
 - $C_2H_4 + [O] \rightarrow CO_2 + H_2O$
 - ozone may also be produced by corona discharge
 - must remove excess O_3 to avoid injury to fruits & vegetables

Source: Applied Ozone Systems

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VI. Avoiding Exposure to Ethylene (cont.)

- Removal of ethylene from storage rooms
 - use of low pressure system (i.e., hypobaric CA storage)
 - $1/10 \text{ atm} = 1/10 \text{ gas concentration in storage atmosphere}$ (e.g., 2% O_2 becomes 0.2% O_2)
 - low pressure also facilitates gas (i.e., C_2H_4) diffusion from fruit tissue

Source: KES Science & Technology, Inc.

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VI. Avoiding Exposure to Ethylene (cont.)

- Removal of ethylene from storage rooms
 - oxidation of ethylene with platinum or oxide catalysts + heat (200-200°C)
 - low temperature catalysis (e.g., $TiO_2 + UV$ radiation at $-100^\circ C$)

Source: KES Science & Technology, Inc.

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VII. Inhibiting Ethylene Biosynthesis & Action

- Biosynthesis inhibition
 - AVG (aminoethoxyvinylglycine)
 - inhibits ACS (i.e., $SAM \rightarrow ACC$)
 - the commercial product is "BeTain"
- Action inhibition
 - 1-MCP (1-methylcyclopropene)
 - Irreversibly binds to ethylene receptors
 - "EthylBloc" for ornamentals and "SmartFresh" for fruits ("Harvista" for preharvest)

Source: Sisler & Serek, 1999, Bot. Bull. Acad. Sin. 40:1-7

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Table 1. Concentration of compound needed to protect plants against ethylene

Compound	Plant	Concentration (nl/l)
2,3,5-SiBP	Banana	50,000
trans-Cyclooctene	Banana	700
cis-Cyclooctene	Banana	512,000
DMCP (darker)	Carrot	700,000
DMCP (lighter)	Carrot	100
1-MCP	Banana	0.7
1-MCP	Carrot	0.5
1-MCP	Pea sprout	20
1,3-BDMCP	Carrot	500

Table 2. Time required for the receptor to become free. 5000% for 1/2 of the receptor to become free after being exposed to the compound. In the case of 1,3-BDMCP and 1-MCP, time refers to time for bananas become sensitive after a single exposure to ethylene.

Compound	Plant	Time (Minutes)
Ethylene	Mang bean sprout	10
Ethylene	Tabacco leaf	10
Ethylene	Tomato leaf	2
2,3,5-SiBP	Mang bean sprout	1800
Trans-cyclooctene	Mang bean sprout	1600
1,3-BDMCP	Banana	20,000
1-MCP	Banana	45,000

Source: Sisler & Serek, 1999, Bot. Bull. Acad. Sin. 40:1-7

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